Palladium-catalyzed coupling reaction of acylzirconocene chlorides with hypervalent iodonium salts: synthesis of aryl-substituted ketones

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Received (in Cambridge, UK) 30th November 2001, Accepted 11th January 2002 First published as an Advance Article on the web 24th January 2002

The palladium-catalyzed acylation reaction of alkenoyland alkanoylzirconocene chlorides with hypervalent iodonium salts afforded the acylated aromatic compounds under mild conditions.

Alkenyl- and alkyl substituted acylzirconocene dichlorides, which are stable and readily available complexes through hydrozirconation of alkynes and alkenes with zirconocene hydrochloride (Cp₂ZrHCl, Schwartz reagent),¹ with subsequent insertion of carbon monoxide² are versatile intermediates as "unmasked" acyl anions for the preparation of aldehydes, carboxylic acids, ketols, and alkylketones as developed by Hanzawa and Taguchi and their co-workers.³ Alkyl-substituted acylzirconium chloride as an "unmasked" acyl anion was coupled with iodobenzene and utilizing the palladium-catalyst under harsh conditions at 100 °C for 20 h gave alkyl phenyl ketone in a low (32%) yield. It is necessary to conduct this coupling under milder conditions and to obtain a better yield. In connection with our program to utilize hypervalent iodonium salts in cross-coupling reactions,⁴ we have found that hypervalent iodonium salts are excellent electrophiles in the palladium-catalyzed coupling reaction of alkenyland alkyl-substituted acylzirconocene chlorides, to form arylsubstituted alkyl and vinyl ketones under mild conditions (Scheme 1).





The results of palladium-catalyzed cross-coupling of the acylzirconocene chlorides with hypervalent iodonium salts are summarized in Table 1.

(E)- β -Styrylacylzirconocene chloride prepared *in situ* from phenylacetylene 1a, Cp₂Zr(H)Cl, and carbon monoxide was reacted in the presence of Pd(PPh₃)₄ (5 mol%) at room temperature for 2 h to afford the α , β -unsaturated aromatic ketone $3a^{5}$ in 75% yield (entry 1 in Table 1). Under the same conditions with (p-methoxyphenyl)(phenyl)iodonium tetrafluoroborate (2b) as an electrophile, the *p*-methoxyphenyl substituted α , β unsaturated ketone $3b^6$ was afforded as the sole product in 73% yield (entry 2). This method was applied to the alkenylsubstituted iodonium salt 2c. β-Styrylacylzirconocene chloride generated in situ was treated with alkenyl-substituted iodonium salt 2c to provide dibenzylideneacetone $3c^7$ in 83% yield (entry 3). Alkyl-substituted acetylene 1b was utilized in the hydrozirconation and carbon monoxide insertion followed by the palladium-catalyzed coupling with 2a, 2b, and 2c to give the coupled products 3d,⁸ 3e,⁹ and 3f¹⁰ in 94, 70, and 75% yields, respectively (entries 4-6). Propargyl alcohol methyl ether (1methoxyprop-2-yne) 1c was converted to the corresponding acylzirconocene chloride and then coupled in the presence of palladium catalyst with 2a, 2b, and 2c to furnish 3g,¹¹ 3h,¹¹ and

DOI: 10.1039/b110983a



In summary, the palladium-catalyzed acylation of alkenoyland alkanoylzirconocene and carbonylation was accomplished with hypervalent iodonium salts to afford acylated aromatic compounds under mild conditions.

Experimental

Typical procedure for the synthesis of 1,3-diphenylprop-2-enone 3a

To a stirred solution of Cp₂Zr(H)Cl (469 mg, 1.82 mmol) in dry THF (10 mL) at room temperature under a nitrogen balloon was added phenylacetylene **1a** (93 mg, 0.91 mmol) and the reaction mixture was stirred for 30 min. After the nitrogen balloon was changed to a CO balloon, the mixture was stirred for 2 h at room temperature. To this solution was added diphenyliodonium tetrafluoroborate **2a** (335 mg, 0.91 mmol) and Pd(PPh₃)₄ (53.2 mg, 0.045 mmol) and the reaction mixture was stirred for 2 h. The reaction mixture was extracted with diethyl ether (20 mL × 3) and the organic layer was dried over MgSO₄, and evaporated *in vacuo*. The crude product was separated by SiO₂ column chromatography (hexanes–EtOAc = 10 : 1, $R_f = 0.35$) to give 1,3-diphenylprop-2-enone **3a** (142 mg, 75%).

Acknowledgements

The authors wish to acknowledge the financial support from the National Research Lab (NRL) program and the Korea Research Foundation (KRF0-2000-015-DP0262). KOSEF-CMDS is gratefully acknowledged. S.-K. Yoon acknowledges financial support from the BK21 program.

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Table 1	The p	alladium-o	catalyzed	coupling	of acylz	irconocene	chlorides	with	hypervalent	iodonium	salts
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Entry	Substrate	Iodonium salts	Products	Isolated yield (%)
1	Рh— <u>—</u> 1а	$Ph_2I^+BF_4^-$ 2a	Ph Ph Ja	75
2	1a	(<i>p</i> -MeO-C ₆ H ₄)I ⁺ PhBF ₄ ⁻ 2b	ph 3b OMe	73
3	1a	$\frac{1}{2c}$	Ph Ph	83
4	<i>п</i> Ви— <u>—</u> 1b	2a	⁰ _n Bu <u>J</u> _{3d}	94
5	1b	2b	nBu 3e OMe	70
6	1b	2c	nBu of ph	75
7	MeO lc	2a	McO Ph	77
8	1c	2b	MeO 3h 0	78
9	1c	2c	MeO Ph	84
10	Ph A	2a	Ph ph Ph	70
11	1d	2b	Ph 3k OMe	69
12	1d	2c	Ph ph ph	72
13	nHex 1e	2a	_{nHex} 3m	68
14	1e	2b	nHex 3n OMe	69
15	1e	2c	nHex Ph	69

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- 129.7, 129.0, 128.4, 125.9, 72.2, 59.5; IR (neat) $\nu = 3057, 2932, 2832, 1725, 1665, 1600, 1447, 1333, 1192, 982, 875, 694 cm⁻¹; HRMS calcd. for C₁₃H₁₄O₂: 202.9994; found: 202.9996.$
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